



CHM 236: Organic Chemistry II

Part two of a two-semester science majors level course on the facts and principles of chemistry as they apply to carbon-based compounds. The course has a mandatory lab that complements the lecture. Topics include reemphasis of lab safety; synthetic techniques; spectroscopy; patterns in the nomenclature, structure, physical properties, spectra; reactivity; stability, stereochemistry and chemical reactions of conjugated systems, benzene and its derivatives, aromatic systems, ethers, carbonyls, amines, carboxylic acids, esters, amides; acid chlorides; anhydrides; nitriles, enols, steroids, lipids, carbohydrates and amino acids; applying the principles of thermodynamics, kinetics and reaction mechanisms to the substitution addition, redox, condensation and elimination reactions of these compounds. Prerequisite(s): CHM 235 with a grade of C or better. Three lecture and three laboratory hours per week. Instructional Support Fee applies.

Course Student Learning Outcomes

- 1. Recognize families of organic compounds based on their functional groups, and apply nomenclature rules to draw formulas, structures, and write names of organic compounds.
- 2. Apply the concepts of isomerism and chirality in organic chemistry, draw Fischer projections, recognize and assessing configurations.
- 3. Apply knowledge of functional group reactivity to propose reasonable mechanisms for basic organic chemistry reactions, and apply knowledge of reaction mechanisms to predict and explain the outcome of a reaction, relative reactivity and stereochemistry.
- 4. Explain the concept of aromaticity and determine if a compound is aromatic, anti-aromatic or non-aromatic.
- 5. Describe the techniques of Nuclear Magnetic Resonance spectrometry, Infrared spectroscopy, and Mass Spectroscopy, and use spectroscopic data to determine molecular structures.
- 6. Plan multi-step synthesis of organic compounds using retrosynthetic analysis and functional group interconversions.

Credits: 4

Program: Chemistry

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